

**AMENDMENTS TO THE CLAIMS**

Please **CANCEL** claims 11 and 12 without prejudice or disclaimer.

Please **AMEND** claim 1 as shown below.

This listing of claims will replace all prior versions, and listings, of claims in the application.

1. (Currently Amended) A lithium secondary battery, comprising:
  - a positive electrode including a material that is capable of reversible intercalation/deintercalation of lithium ions as a positive active material;
  - a negative electrode including a lithium metal, a lithium-containing alloy, a material that is capable of reversible intercalation/deintercalation of lithium ions as a negative active material or a material that is capable of reversibly forming a lithium-containing compound;
  - a separator interposed between the positive and negative electrodes; and
  - an electrolyte on the separator, wherein the electrolyte includes a non-aqueous organic solvent, a lithium salt, and a linear polymer having P=O bonds,
    - wherein the linear polymer having P=O bonds is present in an amount ranging from about 0.005 to less than 5 wt% based on the total amount of the electrolyte,
    - wherein the non-aqueous organic solvent comprises a mixed solvent of a carbonate solvent and an aromatic hydrocarbon solvent, and
    - wherein the aromatic hydrocarbon solvent is at least one selected from the group consisting of fluorobenzene, fluorotoluene, trifluorotoluene, and a compound of Formula (1):



where R is a halogen, or a  $C_1$  to  $C_{10}$  alkyl, and p is an integer of 1 to 6, ~~where p is 3 to 6~~  
 when R is a  $C_4$ -alkyl,

wherein the linear polymer having P=O bonds is formed of a polymerized phosphonate compound with a polymerizable functional unsaturated hydrocarbon group, and

wherein the phosphonate compound with the unsaturated hydrocarbon group is at least one selected from the group consisting of allyl diethylphosphonoacetate

$(C_2H_5O)_2P(O)CH_2CO_2CH_2CH=CH_2$ ), triethyl 3-methyl-4-phosphonocrotonate

$(C_2H_5O)_2P(O)CH_2C(CH_3)=CHCO_2C_2H_5$ ), and allyl tetraisopropylphosphonodiamidite ( $[(CH_3)_2CH]_2Ni_2POCH_2CH=CH_2$ ).

2. (Canceled)

3. (Previously Presented) The lithium secondary battery according to claim 1, wherein the carbonate is selected from the group consisting of dimethyl carbonate (DMC), diethyl carbonate (DEC), dipropyl carbonate (DPC), methylpropyl carbonate (MPC), ethylpropyl carbonate (EPC), methylethyl carbonate (MEC), ethylene carbonate (EC), propylene carbonate (PC), and butylene carbonate (BC).

4. (Previously Presented) The lithium secondary battery according to claim 1, wherein the carbonate solvent comprises a mixed solvent of a cyclic carbonate and a linear carbonate.

5 – 7. (Canceled)

8. (Previously Presented) The lithium secondary battery according to claim 1, wherein the carbonate solvent and the aromatic hydrocarbon solvent are mixed in a volume ratio ranging from about 1:1 to about 30:1 carbonate solvent to aromatic hydrocarbon solvent.

9. (Original) The lithium secondary battery according to claim 1, wherein the lithium salt is at least one selected from the group consisting of  $\text{LiPF}_6$ ,  $\text{LiBF}_4$ ,  $\text{LiSbF}_6$ ,  $\text{LiAsF}_6$ ,  $\text{LiClO}_4$ ,  $\text{LiCF}_3\text{SO}_3$ ,  $\text{Li}(\text{CF}_3\text{SO}_2)_2\text{N}$ ,  $\text{LiC}_4\text{F}_9\text{SO}_3$ ,  $\text{LiSbF}_6$ ,  $\text{LiAlO}_4$ ,  $\text{LiAlCl}_4$ ,  $\text{LiN}(\text{C}_x\text{F}_{2x+1}\text{SO}_2)(\text{C}_y\text{F}_{2y+1}\text{SO}_2)$  (wherein x and y are natural numbers),  $\text{LiCl}$ , and  $\text{LiI}$ .

10. (Original) The lithium secondary battery according to claim 9, wherein the lithium salt is used in a concentration ranging from about 0.6 to about 2.0 M.

11 – 13. (Canceled)

14. (Withdrawn) A method of manufacturing a lithium secondary battery, comprising: preparing a lithium secondary battery by injecting a non-aqueous electrolyte comprising a lithium salt, a non-aqueous organic solvent, a phosphonate compound with an unsaturated hydrocarbon group, and a polymerization initiator between positive and negative electrodes; and reacting the phosphonate compound with the unsaturated hydrocarbon group to prepare a linear polymer with  $\text{P=O}$  bonds.

15. (Withdrawn) The method according to claim 15, wherein the method further comprises heat-treating the manufactured lithium secondary battery.

16. (Withdrawn) The method according to claim 15, wherein the phosphonate compound with the unsaturated hydrocarbon group is at least one selected from the group consisting of diethyl vinyl phosphonate, dimethyl vinyl phosphonate, dipropyl vinyl phosphonate, ethylene glycol methacrylate phosphate ( $\text{CH}_2=\text{C}(\text{CH}_3)\text{CO}_2\text{CH}_2\text{CH}_2\text{OP}(\text{O})(\text{OH})_2$ ), allyl diethylphosphonoacetate ( $\text{C}_2\text{H}_5\text{O})_2\text{P}(\text{O})\text{CH}_2\text{CO}_2\text{CH}_2\text{CH}=\text{CH}_2$ ), triethyl 3-methyl-4-phosphonocrotonate ( $\text{C}_2\text{H}_5\text{O})_2\text{P}(\text{O})\text{CH}_2\text{C}(\text{CH}_3)=\text{CHCO}_2\text{C}_2\text{H}_5$ ), and allyl tetraisopropylphosphorodiamidite ( $[(\text{CH}_3)_2\text{CH}]_2\text{N}]_2\text{POCH}_2\text{CH}=\text{CH}_2$ ).